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covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Chapter 6 Review: Chemical Bonding Flashcards | Quizlet A chemical bond between atoms results from the attraction between electrons and. A shared electron pair. A covalent bond consists of. Nonpolar covalent. If two covalently bonded atoms are

Chapter 6 Chemical Bonding Section 2 Covalent Answer Key

covalent bond. in a covalent bond, the dissociation energy is released in the process of. exothermic reaction. When 2 or more atoms bond by means of electron sharing - the result is. a molecule. shared electrons are centered between the 2 atoms in covalent bonding, the attachment is called. a sigma bond.

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This is the study note for learning all things covalent bonding! Our comprehensive resource covers how covalent bonding works, give examples of covalent bonding and explores the differences between covalent and ionic bonding.

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A covalent bond is the chemical bond formed when two atom share electrons. A molecule is a neutral group of atoms joined by covalent bonds. Know what double and triple bonds are. A double bond is formed when two atoms share two pairs of electrons and a triple bond is formed when atoms share three pairs on electrons.

Study Guide for Chapter 5: Ionic and Covalent Bonds Test ...

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